



## Overcoming Compound Volatility with Sub-ambient ELSD

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### Introduction

The success of evaporative light scattering detection relies on evaporating the eluent without destroying the analyte particle. Where the analyte is non-volatile, evaporator temperatures of 50–100 °C can be used without compound degradation, thus giving maximum sensitivity. However, when compounds have high vapor pressures and the eluent has a high boiling point (e.g. water), detection of semi-volatile compounds is problematic, because the evaporation temperature needs to be set at 30 °C or above in order to evaporate the solvent.

Most of today's ELSDs are only capable of evaporating aqueous eluents at 30 °C, or above, but for low molecular weight, low boiling point compounds even 30 °C is too high, and these compounds will be destroyed in the evaporation process, resulting in poor sensitivity.

To improve the detection of thermally sensitive compounds, evaporation at sub-ambient temperature is required. However, evaporating water at increasingly lower temperatures requires exponentially longer evaporation times. Therefore, temperature alone cannot be used to evaporate water at sub-ambient temperature.

The Varian 385-LC from Polymer Laboratories, now a part of Varian, Inc., has been specifically developed to operate at sub-ambient evaporation temperatures, using a Peltier cooled evaporation tube. The removal of solvent is achieved using a secondary stream of dry nitrogen gas that is added at the evaporation stage. When the evaporation temperature is lowered below ambient temperature, the "evaporation gas" is increased to compensate. At higher evaporation temperatures, the evaporation gas can be reduced or even turned off, to maximize signal-to-noise ratios. Using this sub-ambient technology, eluent can be evaporated as low as 10 °C.

This note demonstrates the value of sub-ambient evaporative light scattering detection in the analysis of semi volatile compounds.

### Conditions (Figure 1)

Sample: Parabens  
Column: Pursuit™ C18 5 µm, 150 x 2.1 mm  
Eluent A: Water  
Eluent B: Acetonitrile  
Gradient: 5–70 % B in 5 min; 70–95 % B in 2 min  
Flow Rate: 0.2 mL/min  
Inj Vol: 10 µL  
Detection: Varian 385-LC (neb=30 °C, evap temp. as shown, gas=1.4 SLM)

### Conditions (Figure 2)

Samples : Acetanilide MW 135  
Column: C18 5 µm, 150 x 2.1mm  
Eluent A: Water + 0.1% TFA  
Eluent B: ACN + 0.1% TFA  
Gradient: 60–90% B in 5 min  
Flow rate: 0.2 mL/min

### Results and Discussion

The benefit of a cooled evaporation zone that provides sub-ambient operation to detect low molecular weight compounds is shown in the analysis of parabens (Figure 1). The lower the temperature, the more parabens detected.

#### Peak identification

1. Methyl paraben
2. Ethyl paraben
3. Propyl paraben
4. Butyl paraben
5. Heptyl paraben

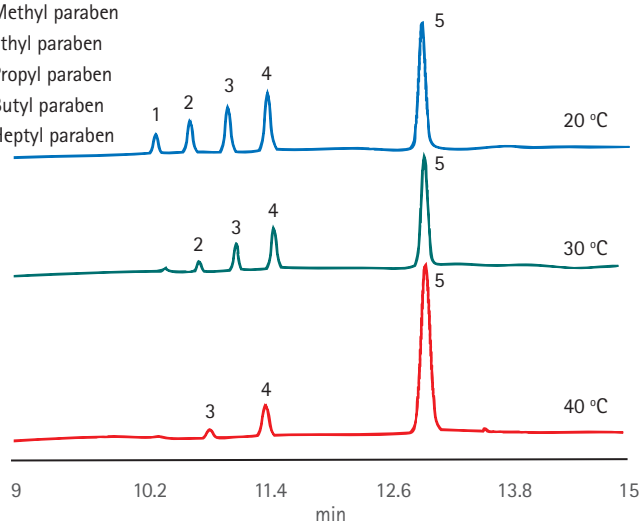


Figure 1. Improved detection of parabens using sub-ambient evaporative light scattering detection (parabens mixture injected at 100 ng except heptyl at 160 ng).

Figure 2 shows the sub-ambient ELS detection of low molecular weight and semi volatile acetanilide, a precursor in penicillin synthesis and other pharmaceuticals including painkillers and intermediates.

In this example the signal-to-noise ratio increases four-fold at 15 °C compared to 30 °C. Sub-ambient ELSD at 20 °C not only increases the number of semi-volatile compounds detectable by ELSD, but also minimizes any response variation between compounds due to volatility differences. Consequently, sub-ambient ELSD can improve the accuracy of compound quantification and uniformity of response.

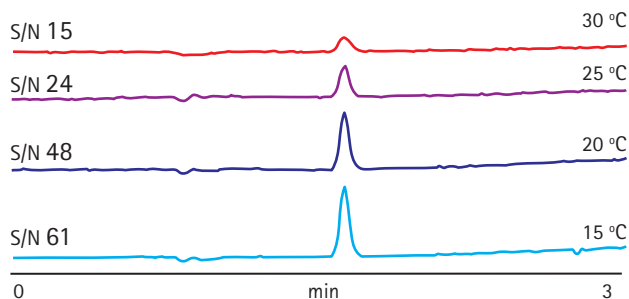


Figure 2. Improved detection of acetanilide at sub-ambient temperatures using evaporative light scattering detection.

## Conclusion

In the past, quantification and purity measurements using evaporative light scattering detection showed poor accuracy due to compound volatility and response variation across a solvent gradient. However, sub-ambient ELSD at 20 °C, as delivered by the Varian 385-LC, increases the accuracy of measurement by minimizing compound volatility effects.

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