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This presentation is used with Lesson Plan 18: Concentration and Unit Conversions in the Drinking Water Quality Testing Trainer Manual.

Available at www.cawst.org/resources

Concentration and Unit Conversions



Learning Expectations

- 1. Explain concentration and units.
- 2. Calculate unit conversions for different concentrations.



Concentration and Units

- Concentration = amount of a component (or chemical) per volume of water, stated in units
- Units = standards of measurement
 - e.g., milligrams, litres
- Without units test values are meaningless
 - e.g., Colony Forming Units/100 mL tell us the number of bacteria in a 100 millilitre volume of water



Parts Per Million (ppm)

- 1 ink drop in 200 L barrel of water = ink concentration of 1 ppm
- 1 ppm = 1 mg/L

 ppm or mg/L on a lab report means the same thing



Parts Per Billion (ppb)

- 1 ink drop in tanker truck used to haul gasoline = ink concentration of 1 ppb
- 1 ppb = 1 μ g/L
- ppb or µg/L on a lab report means the same thing





Unit Conversions Chart

$$1 \text{ ppm} = 1 \text{ mg/L}$$

$$1 \text{ ppb} = 1 \text{ } \mu\text{g/L}$$

$$1 \text{ mg/L} = 1000 \mu g/L$$

$$1 \text{ ppm} = 1000 \text{ ppb}$$

To convert ppb to ppm divide by 1000 To convert µg/L to mg/L divide by 1000

To convert ppm to ppb multiply by 1000 To convert mg/L to µg/L multiply by 1000



Unit Conversion Activity

1. Convert: 0.01 ppm into ppb

Answer: 10 ppb

2. Convert: 20 µg/L into mg/L

Answer: 0.02 mg/L



Unit Conversion Activity

- 3. The WHO Guideline for Drinking Water is <0.01 mg/L for arsenic.
 - You are the project manager for an NGO that has implemented a borehole drilling project.
 - Water quality testing was done on 1 borehole and the lab report states that the arsenic concentration is 8 ppb.
 - Is the water within the WHO Guideline?

Answer: Yes, the results are equal to 0.008 ppm or 0.008 mg/L and are within the WHO Guideline.



Review

True or False

1. Is 1 part per million larger than 1 part per billion?

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True (= 0.000001 is bigger than 0.000000001)
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- 2. 1 ppb = 1 mg/L False (1 ppb = 1 μ g/L)
- 3. $1 \mu g/L = 1 ppb$

